

## **World Ozone Day**

### **“Montreal Protocol@35: Global Cooperation Protecting Life on Earth”**

**Date & Time: 20<sup>th</sup> September 3:00-5:40 pm**

**Start Time: 3:00 pm**

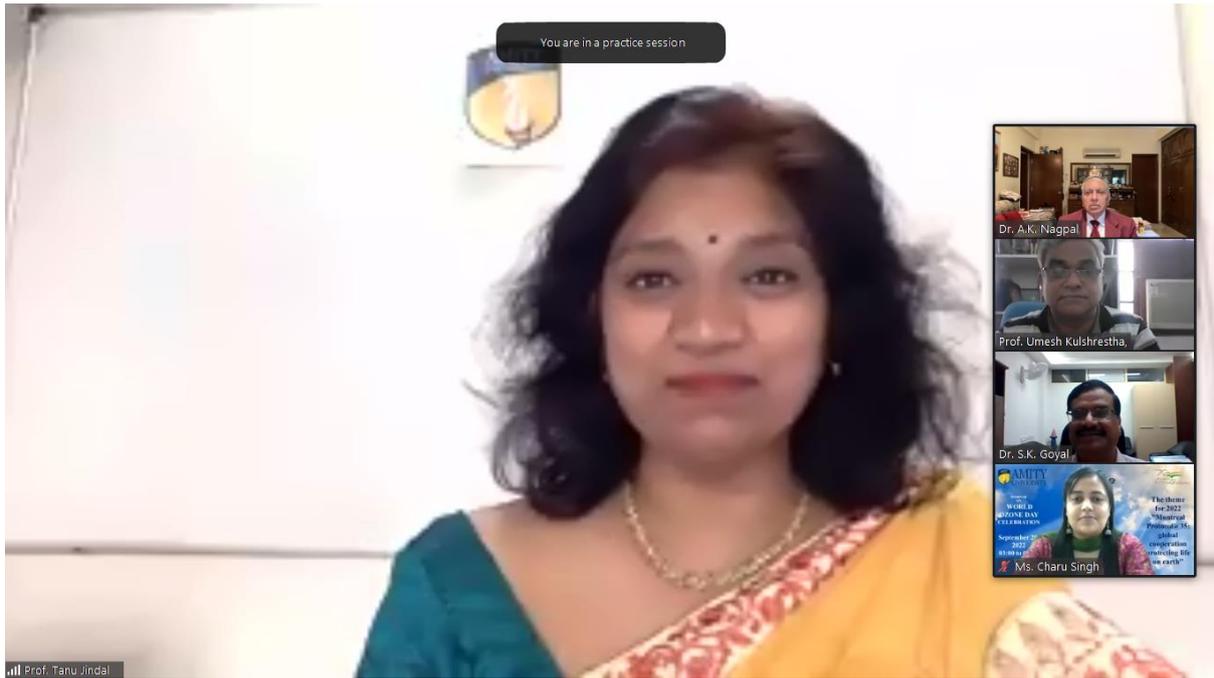
**End Time: 5:40 pm**

**The international webinar was organized on the World Ozone Day. The webinar began at 3:00 pm.**

#### **Introduction to the theme by Prof. Tanu Jindal**

Prof. Tanu Jindal, Director of Amity Institute of Environmental Toxicology, Safety and Management, Amity University, Noida. She welcomed the dignitaries and introduced this year's theme for World Ozone Day. Prof. Tanu Jindal gave a welcome speech with highlight the theme. She discussed her research paper, books, patents, and projects. She said that ozone day is a very important day in terms of UV radiation and the ozone hole. She also discussed the toxic impact of ozone on human health and the environment. She said that UV-induced allergy occurs in 5-20 % of the population often after the first summer exposure to UV radiation. Malignant melanoma, non-melanocytic skin cancer, sunburn, and chronic sun damage are the effects of the skin due to exposure to UV radiation. Acute photokeratitis, cancer of the cornea, lens opacity, uveal melanoma, and macular degeneration are the effects on the eye. The growth of the plants is reduced, and acidity increases in aquatic ecosystems. There are many possible solutions such as reducing the production of those chemicals that cause the destruction of ozone, proper recycling, staying out of the sun, and carbon sequestration.

**“Protect ozone layer save earth to bring worth for the new birth”**



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You are viewing Prof. Tanu Jindal's screen

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**The theme for 2022 is "Montreal Protocol@35: global cooperation protecting life on earth"**

Prof. (Dr.) Tanu Jindal  
Professor and Director  
Group Additional Pro Vice Chancellor (R&D)  
Director  
Amity Institute of Environmental Toxicology, Safety and Management  
Amity Institute of Water Technology and Management  
Amity Centre for Antarctic Research and Studies  
Amity Institute of Oceanography and Atmospheric Sciences  
Advisor, Amity Institute of Marine Science and Technology  
Amity University Uttar Pradesh,  
Sector 125, Noida 201313, India

Prof. Tanu Jindal

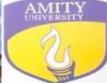
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Participants 44

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### GWCC CLUSTER GROUP

**Group-1: High-resolution climate models** Exploratory studies on climate change, Weather extremes, and climate variability

1. Dr. Pushan Dutta (AUK)
- Dr. PVS Raju (AUR)
2. Dr. Sayantika Mukherjee (AUK)
3. Dr. Abhinav Sahay (AUP)
4. Dr. Abdul Amir Khan (AUH)
5. Dr. Kushagra Rajendra

**Group-2: Biotic and Abiotic Stress tolerance for microbes for sustainable Environment**

1. Dr. Amit C Kharkwal (AIMT, AUUP)
2. Dr. Kuldeep Dwivedi (AUMP)
3. Dr. Richa Sinha (AUJ)
- Dr. Smriti Shukla (AUUP)
4. Dr. Richa Dave Nagar (AUUP)
5. Dr. Richa Sinha (AUJ)

**Group-3: Biofuel, Bioplastics, Biosurfactants, etc for ecofriendly products**

1. Prof. Dr. Indu Shekhar Thakur (JNU)
- Dr. Neeraj Kumar
2. Dr. Abhishek Chauhan (AUUP)
3. Dr. Anuj Ranjan (AUUP)

**Group-4: Bioremediation, Adsorption based technologies for the removal of pollutants from water**

1. Dr. Umesh Kumar (AUP)
2. Dr. Amrita Saha (AUK)
3. Dr. Kuldeep Dwivedi (AUMP)
4. Dr. Anamika Srivastava (AIES, AUUP)
5. Dr. Manoj C. Garg, (AIES, AUUP)
6. Dr. Kartikeya Shukla (AIES, AUUP)

**Group-5: Air pollution (air pollution dispersion modelling and air pollution indexing) Noise Pollution**

1. Dr. Rajshree Sarkar (AUK)
2. Dr. Shubhansh Tiwari (AUH)
3. Dr. Nilesh Deorao Wagh (AUM)
4. Dr. Amrit Kumar (AUH)

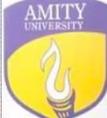
**Group-6: Solid waste, Heavy metal toxicity and removal, plants**

1. Dr. Ashutosh Tripathi (AIES, AUUP)
2. Dr. Ambrina Sardar Khan (AIES, AUUP)
3. Dr. Priyanka Singh (ASET, AUUP)
4. Dr. Renu Dhupper (AIES, AUUP)



Prof. Tanu Jindal

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### AMITY GLOBAL RESEARCH HUB (AGRH), NEW YORK-LONG ISLAND CAMPUS

Amity is in the process of exponential growth in the field of Education and Research. **Dr. Ashok K. Chauhan**, Founder President and Chairman of Amity Group of Institutions & Industries **has envisioned to develop all Amity Universities as Research & Innovation Driven University. To achieve his vision to make India not only a Knowledge Superpower but also a General Superpower by 2030.**

**Dr. Atul Chauhan**, President Ritnand Balved Education Foundation; CEO-AKC Group of Companies and Chancellor Amity University Uttar Pradesh has promoted Research & Innovation in all Amity Institutions.

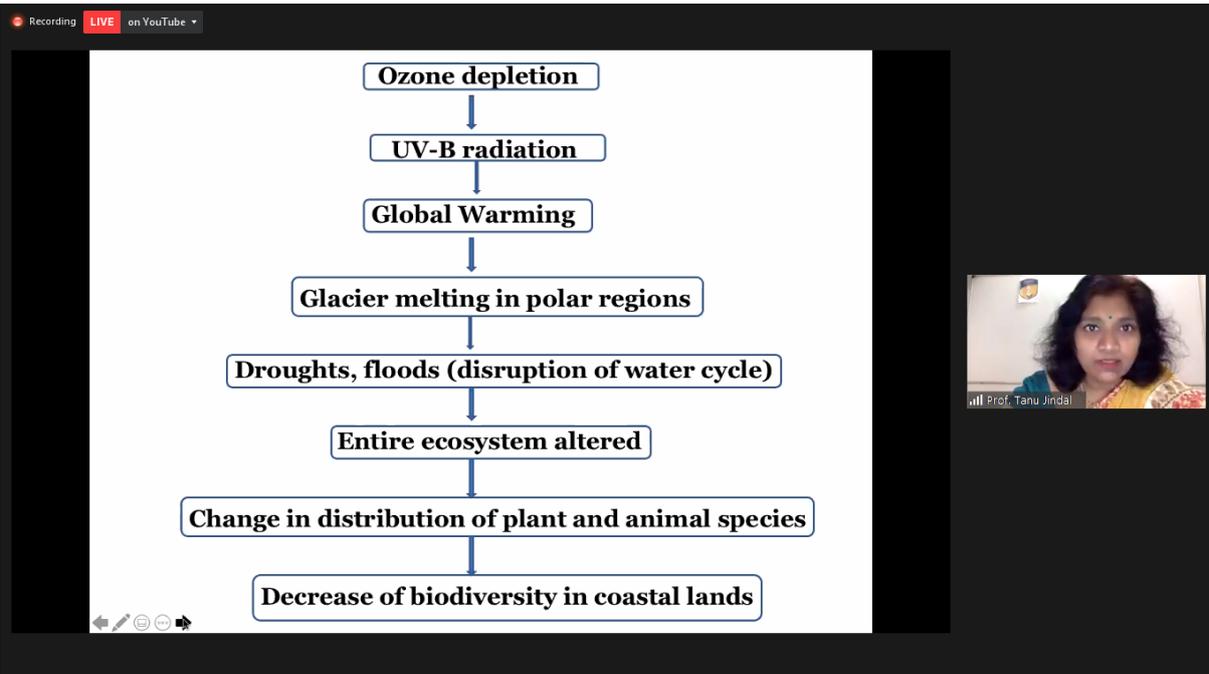
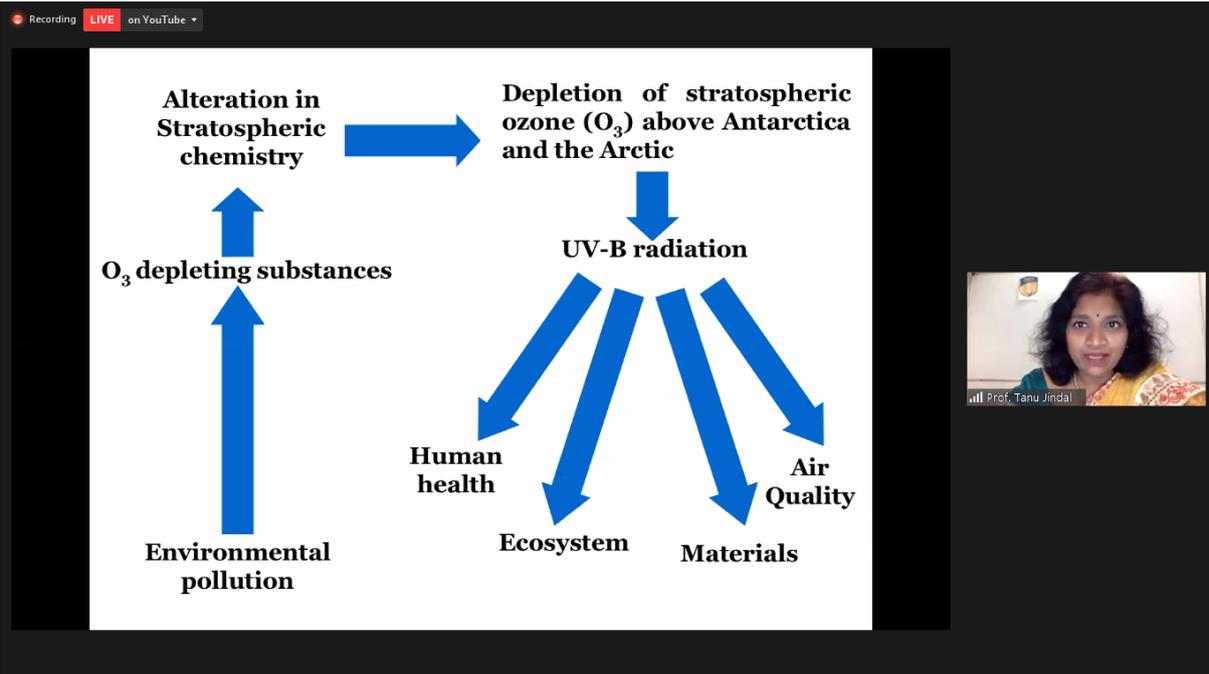
**Dr. Aseem Chauhan**, Additional President, Ritnand Balved Education Foundation; CEO, Amity Capital Ventures and Chancellor, Amity University Haryana **has established an 'Amity Global Research Hub' (AGRH) at Amity New York-Long Island Campus.**







Prof. Tanu Jindal



• **UV-induced allergy**

-Occurs in 5-20% of the population often after first spring/summer exposure to UV radiation



-Increased photosensitivity of the skin (temp., dust - deserts)

-Higher temperatures may lead to more skin cancers  
For the same UV exposure, for every 10°C increase, there is an estimated 3-6% increase in skin cancers



• **Effects on the eye**

- Acute photokeratitis and photoconjunctivitis
- Climatic droplet keratopathy
- Pterygium
- Cancer of the cornea and conjunctiva
- Lens opacity (cataract) – cortical, posterior subcapsular
- Uveal melanoma
- Acute solar retinopathy
- Macular degeneration



**Malignant melanoma of the eye**



**Macular degeneration of the eye**



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## Ozone depletion over India

- With so much worry about the rapid ozone depletion taking place in various parts of the earth, Indian scientists are closely monitoring the ozone layer over India for possible depletion trends
- Since India already receives high doses of ultraviolet (UV-B) radiation, and is at the threshold, effects of ozone layer depletion could be far more disastrous in India
- While there is no trend in the total ozone value, there is some evidence of ozone depletion at higher altitudes- at about 30 to 40 km - even over the tropics

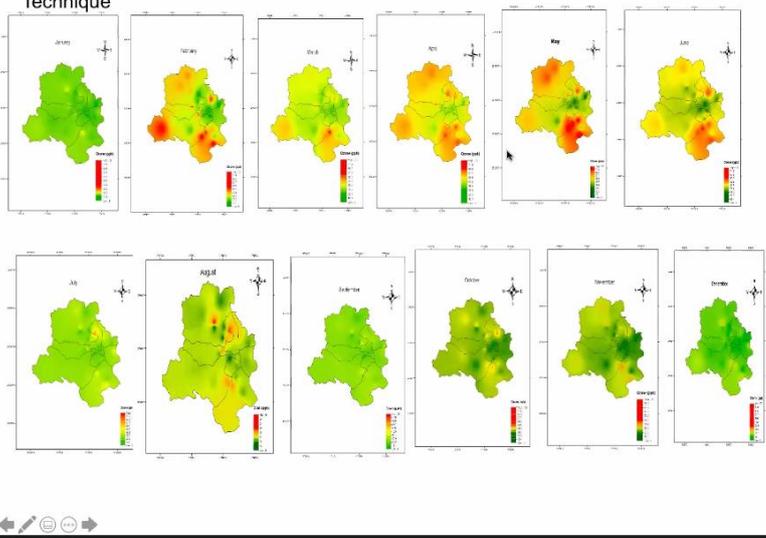


Prof. Tanu Jindal

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### Spatial Variation of Ozone over New Delhi using GIS based Interpolation Technique



Prof. Tanu Jindal

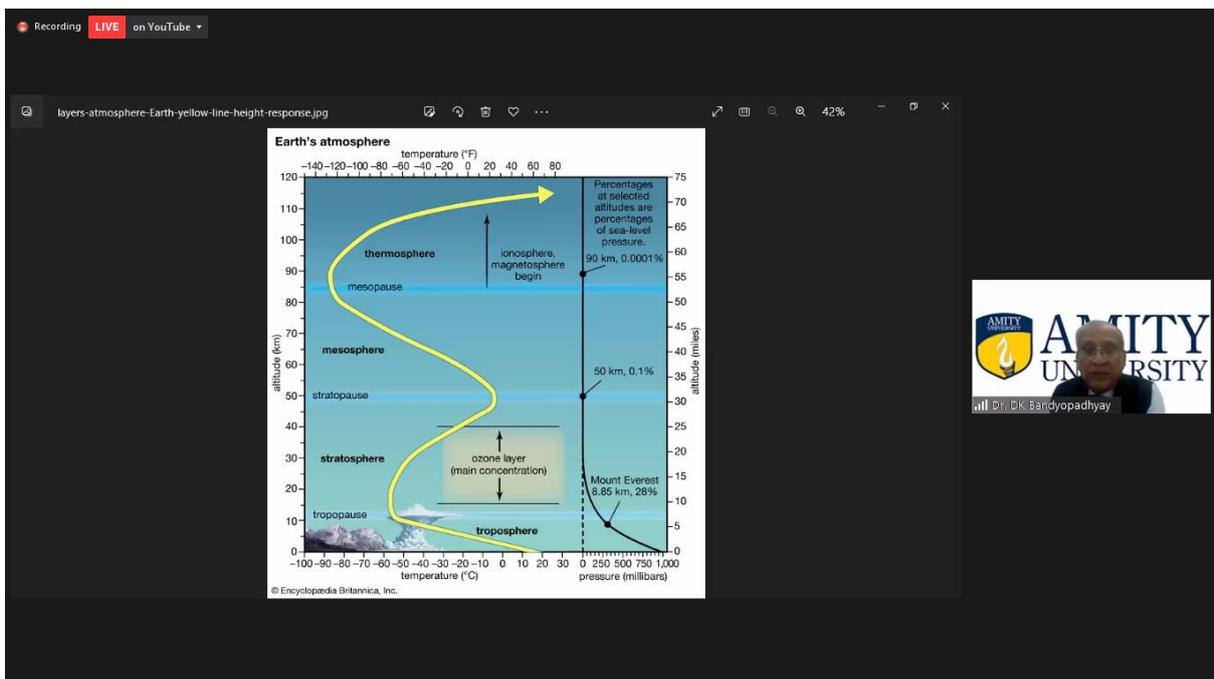
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**Address by Dr. D.K Bandyopadhyay**

**Dr. D.K Bandyopadhyay, Chief Advisor FPO, and Chairman, Amity Law School, Amity University, Noida.** He also welcomed all the dignitaries. He addressed the theme of the ozone day. World Ozone Day is the most important day as per the Montreal Protocol. It is the international treaty which was adopted on 16 September 1987. Initially, it was signed by 46

countries but now it has nearly 200 countries. This treaty regulates the use of chemicals which were dangerous to ozone. CFC was the main responsible compounds for the depletion of ozone. So, many developed countries have taken the decision to ban the use of CFC. People should take the precautions from the UV radiations. Most of the remaining ozone occurs in the troposphere and the layer of the atmosphere is extent from earth to the stratosphere. The harmful UV rays can cause the skin disease.





Presentations were given by-

- 1- **Dr. Mukesh Sharma**, Professor, Department of Civil Engineering, IIT Kanpur.
- 2- **Dr. Vijay Kumar Soni**, Head, Environmental Monitoring and Research Center (EMRC), Delhi.
- 3- **Dr. S. K. Goyal**, Chief Scientist & Head CSIR-NEERI, Delhi.

- 4- **Prof. Manju Mohan**, Head, Centre for Atmospheric Sciences, Indian Institute of Technology, New Delhi.
- 5- **Prof. Umesh Kulshrestha**, Deputy Director, School of Environmental Sciences, Jawaharlal Nehru University, New Delhi.
- 6- **Prof. Qing-Bin Lu**, Professor, University of Waterloo, Ontario, Canada.

**Speaker-wise discussion points (as per programme flow):**

**1. Dr. Mukesh Sharma- “Ozone: Basic to its stratospheric resurrection”**

- 1.1. Ozone is a highly reactive gas composed of three oxygen atoms.
- 1.2. It is both a natural and man-made product that occurs in the Earth’s upper atmosphere (stratosphere) and lower atmosphere (troposphere).
- 1.3. It works a lot like sunscreen, blocking out harmful UV rays from the sun.
- 1.4. The ozone layer is getting depleted due to anthropogenic pollution forming an ozone hole.
- 1.5. Since the 1970’s the ozone hole has been increasing in size over Antarctica.
- 1.6. For the first time, in September of 2000, the ozone hole became so large it actually left populated areas of Southern Chile fully exposed to the effects of the Sun’s UV rays.

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**Dr. Mukesh Sharma**  
Professor  
Department of Civil  
Engineering, IIT Kanpur

He is a renowned professor in department of civil engineering IIT Kanpur. He has completed PhD in civil engineering from University of Waterloo, Ontario, Canada, and MTech in civil engineering from IIT Kanpur. Dr. Sharma holds honorary positions like Vice President of Indian Air Pollution Control Association, Delhi. He has also served in Central Pollution Control Board, delhi for >10 years. He has received many awards and recognition such as Fellow of National Academy of Engineering (FNAE) 2014, Kong Ha Award in Air Quality from Clean Air Asia (a UN supported foundation) 2012, Hiyoshi Environmental Award by Hiyoshi Corporation, Japan, Excellence in Research and Technology Development, 2012, etc. He has published more than 86 papers in reputed journals.



The theme for 2022 "Sustainable Planet 35: global cooperation unceasing life on earth"

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Dr. DK Bandyopadhyay



Ms. Jayati Arora



Ms. Charu Singh

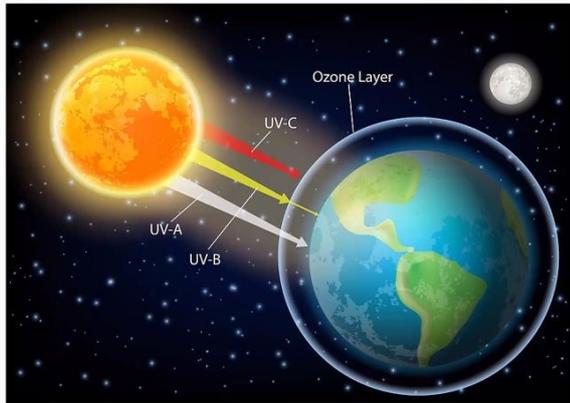


Prof. Tanu Jindal



Dr. Mukesh Sharma

- The ozone layer is a protective layer of gas molecules located within the stratosphere.
- It Acts as an umbrella or shield of the earth to protect it from incoming harmful UV rays from sun.



## The Ozone Hole

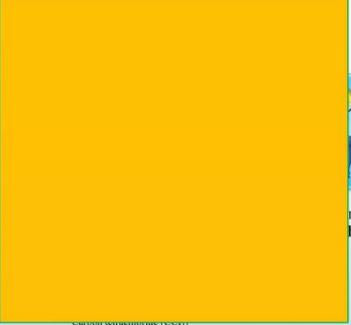
### Causes of Ozone Depletion

- Natural Factors
  - ✓ Seasons
  - ✓ Weather Conditions
  - ✓ Solar Cycle
- Anthropogenic
  - Ozone layer is being destroyed by a group of manufactured chemicals that are called **ODS** or **Ozone-Depleting Substances**
    - ✓ Chlorofluorocarbons (CFC 11, CFC 12, CFC 13)
    - ✓ Methyl bromide (CH<sub>3</sub>Br)
    - ✓ Halons (halon-1211, halon-1301, halon-2402)
    - ✓ Carbon tetrachloride (CCl<sub>4</sub>)
    - ✓ Hydrochlorofluorocarbon (HCFC 22 and HCFC 123)



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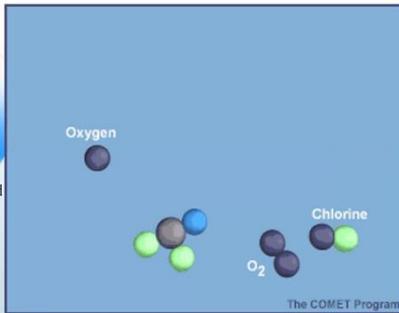
## The Ozone Hole



manufactured substances

Carbon tetrachloride (CCl<sub>4</sub>)

Hydrochlorofluorocarbon (HCFC 22 and HCFC 123)



Oxygen

Chlorine

O<sub>2</sub>

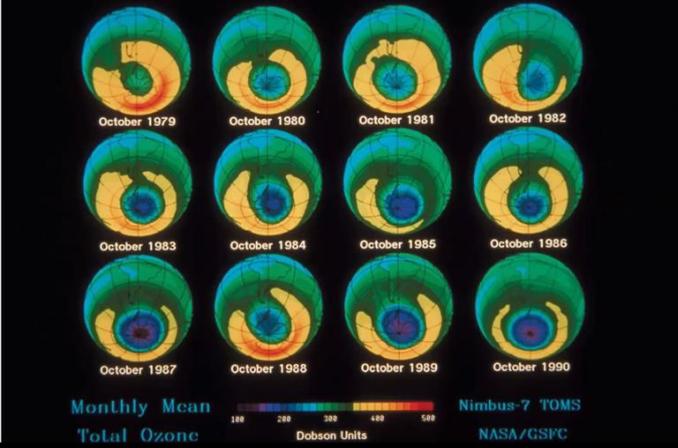
The COMET Program



Dr. Mukesh Sharma

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## Increase of Ozone Hole over the south pole from 1979-1990



October 1979    October 1980    October 1981    October 1982

October 1983    October 1984    October 1985    October 1986

October 1987    October 1988    October 1989    October 1990

Monthly Mean Total Ozone    100    200    300    400    500    Dobson Units    Nimbus-7 TOMS    NASA/GSFC



Dr. Mukesh Sharma

## The Montreal Protocol -1987



Dr. Mukesh Sharma

## Montreal 2007 adjustment: HCFC phase-out

- September 21, 2007 in Montreal:**
- Adjustment of Montreal Protocol: accelerated HCFC phase-out
  - Climate effects taken into account
  
  - Developed countries:
    - Phase-out from 2030 → 2020 (+ intermediate reductions targets)
  - Developing countries:
    - Freeze in 2012
    - Phase-out from 2040 → 2030 (+ intermediate reductions targets)
    - Base level from 2015 → average 2009-2010



Dr. Mukesh Sharma

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### How the Global community and EU have controlled the emission of the ozone depleting chemicals

Consumption in Ozone-Depleting Potential (ODP) tonnes, 1986 = 100

Legend: EU-27 + UK, Global

\* source: European Environment Agency

Dr. Mukesh Sharma

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### Effect on ozone layer

- Mid-latitude: EESC back to 1980-levels around 2050
- Polar region: EESC back to 1980-levels around 2065:
  - Older age of air in polar vortex
- Large ozone depletion without Montreal Protocol and amendments

Effective Stratospheric Chlorine

Legend: No Molina & Rowland, Montreal 1987, London 1990, Copenhagen 1992, Beijing 1999

Y-axis: EESC (ppb)

X-axis: Year

13

Glenn Felders, Montreal Protocol and Climate

Dr. Mukesh Sharma

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The screenshot shows a Zoom meeting interface. At the top, it says 'Recording LIVE on YouTube' and 'You are viewing Dr. Mukesh Sharma's screen'. The main content is a presentation slide titled 'How we are resurrecting'. The slide features two globes representing Earth in the year 2056. The left globe is labeled 'With Montreal Protocol' and shows a green and yellow color gradient, indicating higher ozone levels. The right globe is labeled 'Without Montreal Protocol' and shows a blue color gradient, indicating lower ozone levels. Below the globes is a color scale for 'Ozone Concentration' in 'Dobson Units', ranging from 0 to 600. The video feed on the right shows Dr. Mukesh Sharma, a man with glasses wearing a light blue shirt. The Zoom control bar at the bottom includes options like 'Join Audio', 'Start Video', 'Participants' (68), 'Q&A', 'Chat', 'Share Screen', 'Raise Hand', 'Record', 'Apps', and a 'Leave' button.

## 2. Dr. Vijay Kumar Soni-“Ozone Layer Hole in a Changing Climate”

- 2.1. Ozone or trioxxygen is an allotrope of oxygen that is much less stable than the diatomic allotrope.
- 2.2. Oxygen is a colourless, odourless gas at ambient concentrations.
- 2.3. At high concentration, it is a pale blue gas, slightly soluble in water and much more soluble in inert non-polar solvents. At  $-112\text{ }^{\circ}\text{C}$  temperature, it condenses to form a dark blue liquid.
- 2.4. Ozone is present in low concentrations throughout the Earth’s atmosphere. In total, ozone makes up only 0.6 ppm of the atmosphere.
- 2.5. Ozone is a major product of air pollution and globally its abundance is unknown.
- 2.6. Ozone at the earth’s surface plays a key role in the chemical cycling of many other trace gases in the atmosphere.
- 2.7. Knowing the amount of ozone helps assess the degree to which it is an environmental problem.
- 2.8. The clearest sign, so far, are ozone increases over the last 20 years in the upper stratosphere, and a decrease in the severity of the Antarctic ozone hole in September.

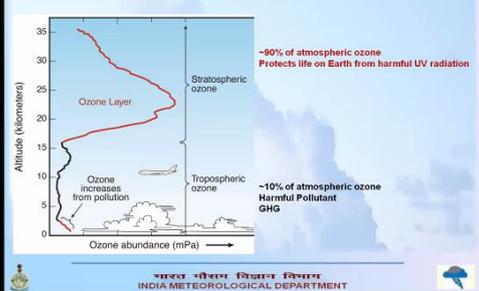
2.9. It is certain that the significant ozone decline from the 1960s to the 1990s has levelled off.

2.10. The worldwide ban of ozone-depleting substances (ODS) by the Montreal Protocol and its amendments has been successful.

The screenshot shows a live YouTube recording. On the left, a presentation slide is displayed with the following text: "Ozone Layer Hole in a Changing Climate", "2022 World Ozone Day", "Montreal Protocol@35: global cooperation protecting life on earth", and contact information for Vijay Kumar Soni at the Environment Monitoring & Research Centre, India Meteorological Department. On the right, a video feed shows Dr. Vijay Kumar Soni speaking.

The screenshot shows a live YouTube recording. On the left, a graph titled "OZONE 'Can't live with it, can't live without it'" plots ozone abundance (mPa) against altitude (kilometers). The graph shows a peak in the stratosphere and a rise in the troposphere. Text on the graph states: "Stratospheric ozone -90% of atmospheric ozone Protects life on Earth from harmful UV radiation" and "Tropospheric ozone -10% of atmospheric ozone Harmful Pollutant GHG". On the right, a video feed shows Dr. Vijay Kumar Soni speaking.

# OZONE "Can't live with it, can't live without it"



This equilibrium is disturbed by the oxidation of NO to NO<sub>2</sub> by peroxy radicals (HO<sub>2</sub>) formed in the course of oxidation of carbon monoxide (CO) and hydrocarbons by OH radicals:

For CO:

$$\text{CO} + \text{OH} \longrightarrow \text{CO}_2 + \text{H}$$

$$\text{H} + \text{O}_2 \longrightarrow \text{HO}_2$$

$$\text{HO}_2 + \text{NO} \longrightarrow \text{OH} + \text{NO}_2$$

For hydrocarbons (e.g., CH<sub>4</sub>):

$$\text{CH}_4 + \text{OH} \longrightarrow \text{CH}_3 + \text{H}_2\text{O}$$

$$\text{CH}_3 + \text{O}_2 \longrightarrow \text{CH}_3\text{O}_2$$

$$\text{CH}_3\text{O}_2 + \text{NO} \longrightarrow \text{CH}_3\text{O} + \text{NO}_2$$

$$\text{CH}_3\text{O} + \text{O}_2 \longrightarrow \text{CH}_2\text{O} + \text{HO}_2$$

$$\text{HO}_2 + \text{NO} \longrightarrow \text{OH} + \text{NO}_2$$

The net result of this chemistry is to produce NO<sub>2</sub> from NO by other means than by reaction (3), thus leading to enhanced ozone concentrations following reactions (1) and (2).

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INDIA METEOROLOGICAL DEPARTMENT



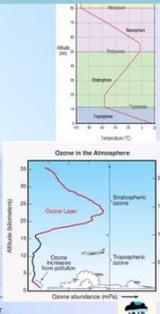
### Variation of Ozone with Altitude

The Ozone layer can be divided into two main parts :

- ❖ Tropospheric Ozone
  - Surface Ozone
  - Upper Tropospheric Ozone
- ❖ Stratospheric Ozone



The radiosonde package collects measurements of ozone, temperature, humidity, pressure, and GPS data once per second throughout the 2.5 hour flight.



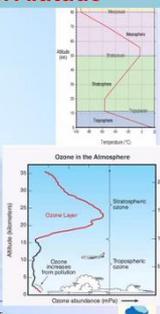
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INDIA METEOROLOGICAL DEPARTMENT



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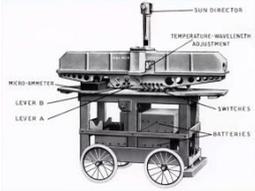
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## Columnar Ozone Measurement

1. Dobson Ozone Spectrophotometer
2. Brewer spectrophotometers




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Dr. Vijay Kumar Soni

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## Stratospheric Ozone Production

Stratospheric Ozone Production

Step 1:  $O_2 + \text{Ultraviolet Sunlight} \rightarrow 2O$

Step 2:  $O + O_2 \rightarrow O_3$

Overall reaction:  $3 O_2 \xrightarrow{\text{sunlight}} 2 O_3$



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Dr. Vijay Kumar Soni

### What does the ozone layer do for us? Ozone Absorption in the UV Band

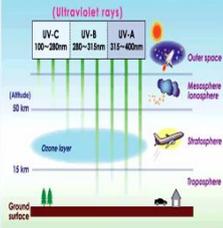
❖ UV radiation includes wavelengths from 200 to 400 nm

- ❖ UV-A 315 ~ 400 nm
- ❖ UV-B 280 ~ 315 nm
- ❖ UV-C 100 ~ 280 nm

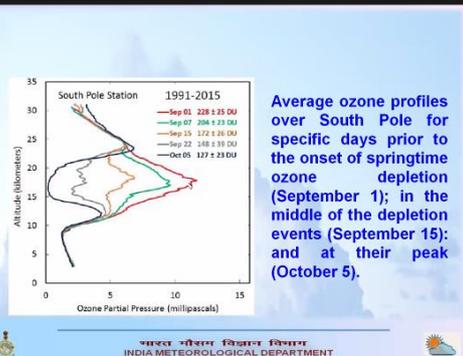
❖ UV-C  
• Nearly all UV-C is absorbed in the upper atmosphere

❖ UV-B  
• 90% of UV-B is absorbed by the atmosphere, mostly by O<sub>3</sub>

❖ UV-A  
• Not strongly absorbed by the atmosphere



Dr. Vijay Kumar Soni



Dr. Vijay Kumar Soni

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### Detection of the Antarctic Ozone Hole

Crosses are British Antarctic Survey (BAS) measurements, triangles and circles are NASA satellite measurements. Measurements are October averages. BAS reported their findings in 1985. NASA later verified their results. Farman et al. publish their famous paper in Nature in 1985

Dr. Vijay Kumar Soni

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### Seasonality of the Antarctic Ozone Hole

Summer Mar 20 Polar Vortex Develops  
Polar Stratospheric Clouds Form  
Summer Sep 23 Ozone Hole Peaks

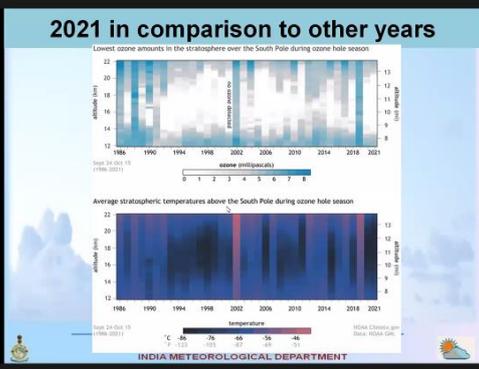
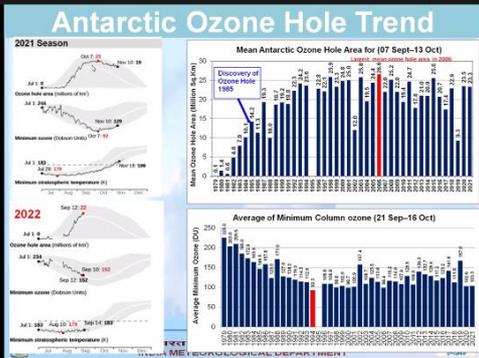
Temperature (C)

Total Ozone (DU)

Altitude (km)

Climate Variability & Predictability - World Climate Research Programme (www.civr.org)

Dr. Vijay Kumar Soni



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**Institutional Frame Work for Implementation of Montreal Protocol in India**

- Empowered Steering Committee**
  - Chaired By: Secretary, EF & CC
  - Members: Secretaries of Related Ministries ( Finance, Industries, Chemicals etc.) / Nominated Experts
  - Mandate: Policies Action Pertaining to Montreal Protocol.
- Montreal Protocol Parties Multilateral Fund Ozone**
  - Ministry of Environment, Forest & Climate Change (MOEF & CC)
  - OZONE CELL**
    - Mandate: Management/Coordination of all MP Activities
- Standing Committee for Technology & Finance**
  - Members: Experts from Industry & Government Institutions
  - Mandate: Review /Endorse MP Activities
- Standing Committee for Monitoring & Evaluation**
  - Members: Experts/Officials from Government & Industry
  - Mandate: Advice on monitoring & Evaluating Activities.
- Project Management Unit For Phase-out of ODS**
  - Monitoring of implementation of ODS Phase-out Projects and activities, Management Information Systems (MIS), Technical Audit, Awareness & Capacity-Building

INDIA METEOROLOGICAL DEPARTMENT

Dr. Vijay Kumar Soni

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With continued production of CFCs, ozone levels worldwide would have dropped to dangerously low levels.

Source: (NASA images by the GSFC Scientific Visualization Studio)

Dr. Vijay Kumar Soni

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### Indian Antarctica Stations



Bharati

Maitri research station

भारती मैत्री भारत भारत  
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Dr. Vijay Kumar Soni

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### Observations at Bharati Antarctica (Indian Station)



Dr. Vijay Kumar Soni

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### Total ozone columns over Bharati

Dr. V. Vijay Kumar Soni

### 3. Dr. S. K. Goyal- “Ozone Layer: Protect the Protector”

- 3.1. In 1985, using satellites, balloons, and surface stations, a team of researchers discovered a balding patch of ozone in the upper stratosphere, the size of the United States, over Antarctica.
- 3.2. The ozone hole is defined as the area having less than 220 Dobson units (DU) of ozone in the overhead column (i.e., between the ground and space).
- 3.3. Composed of elements chlorine, fluorine, and carbon.
- 3.4. CFC was developed in 1903 by DuPont.
- 3.5. CFCs were welcomed by industries: low toxicity, chemical stability, and cheap.
- 3.6. CFCs are used as refrigerants, blowing agents, for making flexible foam, cleaning agents, propellants.
- 3.7. A panel of experts was formed to investigate substances responsible for hole formation.
- 3.8. Established policies that prevent future use of certain types of chemicals.

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**Dr. S.K. Goyal**  
Chief Scientist & Head  
CSIR-NEERI Delhi Zonal  
Centre, Naraina, New Delhi

Dr. S.K. Goyal is Chief Scientist & Head, CSIR-NEERI, Delhi Zonal Center . He has done masters in Chemical Engineering with specialization in industrial pollution abatement from IIT Roorkee in the year 1989 and PhD in Environmental Science & Technology from Nagpur University in 2002. He has more than 100 research publications in journals/conference and supervised about 20 MSc/M.Tech students on air and water related topics. He received Prof. R.C. Singh Prize for Greatest Practical Importance Paper by the Institution of Engineers (India) in December 2006 for the paper "Optimization of Aerated Lagoon System Treating Municipal Wastewater" for presenting his talk on "Ozone for Life : An Indian Prospective".



Dr. S.K. Goyal

3.9.

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"Webinar on Ozone Day Celebration 2022"

**Ozone Layer: Protect the Protector**





Organized by  
Amity University, Noida  
September 29, 2022  
(3-5 pm)



**Dr. S. K. Goyal**  
Chief Scientist & Head,  
CSIR-NEERI Delhi Zonal Centre,  
Naraina, New Delhi

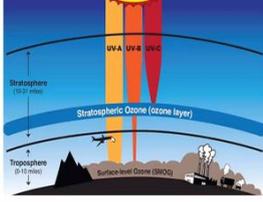


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### Our Atmosphere & Ozone Layer

(N<sub>2</sub>, O<sub>2</sub>, water vapour, CO<sub>2</sub>, etc.)

**Formation of Ozone Layer**  
 $O_2 + \text{sunlight} \rightarrow O + O$   
 $O + O_2 \rightarrow O_3$

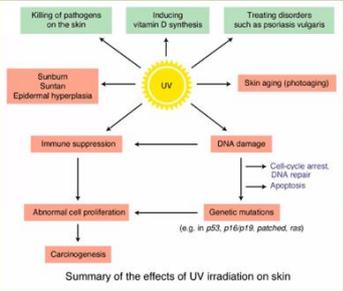


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### What if Ozone Layer is not there - Effects of UV Rays



Summary of the effects of UV irradiation on skin



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### The Discovery of Ozone Hole

- In 1985, using satellites, balloons, and surface stations, a team of researchers had discovered a balding patch of ozone in the upper stratosphere, the size of the United States, over Antarctica.





British Antarctic Survey Research station, Holly Bay, Antarctic coast

Team who discovered the hole 1985.  
From left: Joe Farman, Brian Gardiner, and Jonathan Shanklin

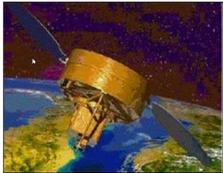


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### Measurement of Stratospheric Ozone

- Total Ozone Mapping Spectrometer (TOMS) used by NASA to measure ozone concentrations
- TOMS – a satellite-borne instrument
- TOMS launched in 1996 – makes 35 measurements every 8 seconds



Artist's view of the QuikTOMS spacecraft (image credit: NASA)

- Levels of ozone are measured in Dobson units (DU), where 100 DU is equivalent to a 1 mm thick layer of pure ozone



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### Mechanism of Ozone Depletion

Ozone layer absorbs 93 % of the harmful, high frequency Ultra-violet radiations.

- 1 - CFCs released
- 2 - CFCs rise into ozone layer
- 3 - UV releases Cl from CFCs
- 4 - Cl destroys ozone
- 5 - Depleted ozone → more UV
- 6 - More UV → more skin cancer

Cl atoms from CFCs attack the ozone, taking away ozone and forming chlorine monoxide (ClO), which then combines with another oxygen atom to form a new oxygen molecule and a chlorine atom.

$$\text{O}_3 + \text{Cl} \rightarrow \text{O}_2 + \text{ClO}$$

$$\text{ClO} + \text{O} \rightarrow \text{Cl} + \text{O}_2$$

**The chlorine atom is free to destroy up to 100,000 ozone molecules**

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### Stratospheric Chlorine Concentration resulted from Different ODS

Stratospheric chlorine (ppb)

Year

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**What about Ground Level Ozone**

Stratospheric O<sub>3</sub>

VOCs, CO + NO + NO<sub>2</sub> + HO, HO<sub>2</sub> → O<sub>3</sub>

Anthropogenic and biogenic emissions → PM<sub>2.5</sub> → Ozone Formation

Ground Level Ozone Formation: Sunlight, Nitrogen oxides, Volatile Organic Compounds, Mobile and Combustion Sources, Vegetation, Mobile Sources and Industry, Pollutants "bake" together in direct sunlight forming ozone.

- Ozone in the stratosphere can move downward to the troposphere, contributing to the "background" level of ground-level ozone. However, high levels of ozone in the troposphere are due to photochemical reactions involving VOCs and oxides of nitrogen (NOx: NO, and NO<sub>2</sub>).
- Anthropogenic emissions (e.g., fossil fuel combustion) are responsible for NOx and mainly responsible for VOCs and CO. Trees also emit certain VOCs (e.g., isoprene).
- PM<sub>2.5</sub> from primary emission sources can react with (consume) free radicals (e.g., HO<sub>2</sub>) responsible for ozone formation, which partly explains the observations in certain areas where ozone level increased while PM<sub>2.5</sub> level decreased. hv, photon; VOCs, CO, NO, NO<sub>2</sub>, NOx; NO and NO<sub>2</sub>; HO, the hydroxyl radical; HO<sub>2</sub>, hydroperoxy radical; PM<sub>2.5</sub>.

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**Ground Level Ozone and NO<sub>2</sub>**

**Standards**

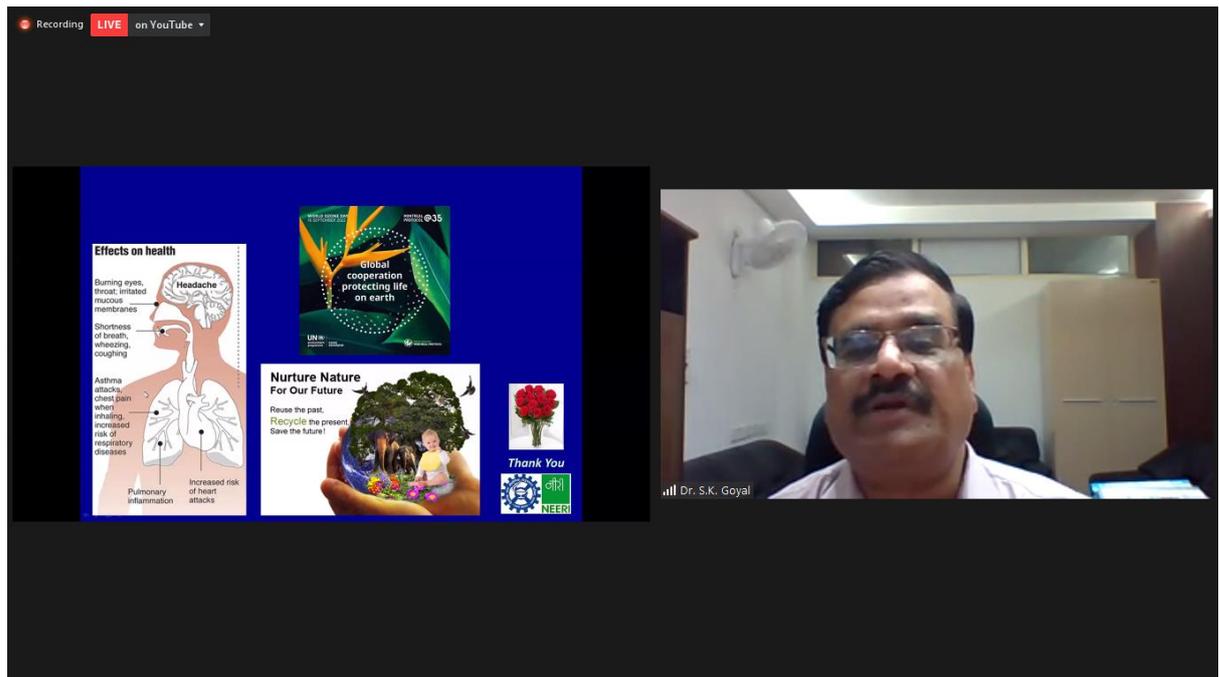
**Ozone**  
1 hr: 180 ug/m<sup>3</sup>  
8 hrs: 100 ug/m<sup>3</sup>

**NO<sub>2</sub>**  
24 hrs: 80 ug/m<sup>3</sup>

CRRRI - Oct 2020

Dr. S.K. Goyal

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#### 4. Prof. Manju Mohan- “For Life on Earth Protect the Ozone Layer and Prevent Climate Change”

4.1 197 countries agreed to the Montreal Protocol.

4.2 To protect the ozone layer by taking measures to control total global production and consumption of substances that deplete it.

4.3 Stratospheric ozone is present in high altitudes in the atmosphere and protects life on earth from damaging UV radiation from the sun.

4.4 Tropospheric ozone forms near the ground from pollutants in presence of sunlight. It is the primary component of smog.

4.5 Human activities such as driving cars and generating electricity are major sources of tropospheric ozone.

4.6 Arctic stratosphere has cooled in recent years, and a strong correlation is found between this cooling and increased O<sub>3</sub> depletion.

4.7 A growth of hydrofluorocarbons (HFCs) in air conditioning and refrigeration systems is seen as a result of CFCs phase-out.

4.8 Photochemical production of ozone is a highly nonlinear system.

4.9 High VOC/NO<sub>x</sub> ratios are less sensitive to changes in VOC concentrations.

4.10 O<sub>3</sub> increases with increasing VOC concentrations and decreases with increasing NO<sub>x</sub> concentrations.

The screenshot shows a Zoom meeting interface. At the top, it says "Recording LIVE on YouTube". The main content area is split into two panels. The left panel features a slide with the AMITY UNIVERSITY logo and a 75th Azadi Ka Amrit Mahotsav logo. The slide includes a portrait of Professor Manju Mohan and text describing her as a professor in the Centre for Atmospheric Sciences [CAS], Indian Institute of Technology [IIT] Delhi. The right panel shows a video feed of Ms. Charu Singh. To her right is a graphic for the "WORLD OZONE DAY CELEBRATION" on September 20, 2022, from 03:00 to 05:40 PM. The theme for 2022 is "Montreal Protocol@35: global cooperation protecting life on earth".

This screenshot shows a Zoom meeting interface with a different view. At the top, it says "Recording LIVE on YouTube" and "You are viewing Prof. Manju Mohan's screen". The main content area is split into two panels. The left panel shows a slide titled "Ozone Day" with the text "FOR LIFE ON EARTH PROTECT THE OZONE LAYER AND PREVENT CLIMATE CHANGE" and "WORLD OZONE DAY". The right panel shows a video feed of Prof. Manju Mohan. The Zoom control bar at the bottom includes options for "Join Audio", "Start Video", "Participants" (70), "Q&A", "Chat", "Share Screen", "Raise Hand", "Record", "Apps", and a "Leave" button.

### Stratospheric v/s Tropospheric Ozone

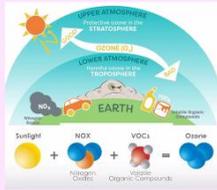
Ozone is Good Up High and Bad Nearby

#### Stratospheric Ozone

- present in high altitudes in the atmosphere
- protects life on earth from damaging Ultraviolet radiation from the sun.

#### Tropospheric Ozone

- forms near the ground from pollutants in presence of sunlight.
- primary component of smog.
- human activities such as driving cars and generating electricity are major sources



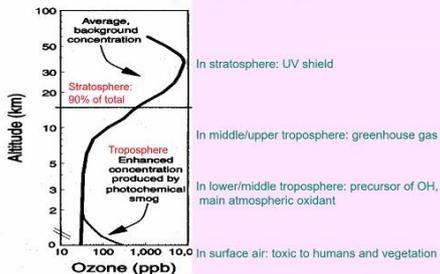
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IIT Delhi

Ozone Day Webinar On September 20, 2022  
Amity Institute of Environmental Toxicology, Safety and Management



Prof. Manju Mohan

### The many faces of atmospheric ozone: Radiation and Chemistry both plays an important role



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### Antarctica Ozone Hole (2000-2019): loss during spring and again make it up

2000      2005      2010

Source: <https://earthobservatory.nasa.gov/world-of-change/Ozone>

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Meeting Information

### Ozone Hole prior to 2000 Over Antarctica

September 1980      September 1985  
September 1990      September 1995

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### PSC FORMATION AT COLD TEMPERATURES

#### Minimum Air Temperatures in the Polar Lower Stratosphere

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### Radiative forcing - Tropospheric Ozone

RF Terms	RF values (W m <sup>-2</sup> )	Spatial scale	LOStu
Long-lived greenhouse gases	CO <sub>2</sub> : 1.66 [1.49 to 1.83] CH <sub>4</sub> : 0.48 [0.42 to 0.53] N <sub>2</sub> O: 0.16 [0.14 to 0.18] Halogencarbon: 0.21 [0.17 to 0.25]	Global	High
Ozone	Stratospheric: -0.05 [-0.15 to 0.05] Tropospheric: 0.35 [0.25 to 0.45]	Continental to global	Med.
Stratospheric water vapor from CH <sub>4</sub>	0.07 [0.02 to 0.12]	Global	Low
Surface albedo	Land use: -0.2 [-0.4 to 0] Black carbon on snow: 0.1 [0 to 0.2]	Local to continental	Med - Low
Direct effect	-0.5 [-0.9 to -0.1]	Continental to global	Med - Low
Total Aerosol	-0.7 [-1.8 to -0.3]	Continental to global	Low
Cloud albedo effect	-0.7 [-1.8 to -0.3]	Continental to global	Low
Linear contrasts	0.01 [0.003 to 0.02]	Continental	Low
Natural	Solar irradiance: 0.12 [0.06 to 0.20]	Global	Low
Total net anthropogenic	1.6 [0.6 to 2.4]		

Tropospheric Ozone is the third largest greenhouse gas contributor to radiative forcing of climate change:  $0.35 \text{ Wm}^{-2}$  ( $\text{CO}_2$ :  $1.66 \text{ Wm}^{-2}$ ;  $\text{CH}_4$ :  $0.48 \text{ Wm}^{-2}$ )

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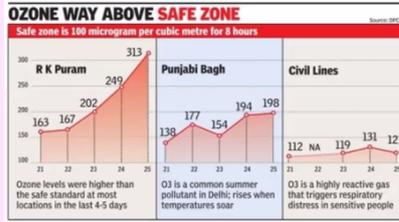
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## Ozone Pollution Worsens as Daily Temperatures Increases

Heat pulls city out of comfort ozone



Source: TOI, 26 May, 2016



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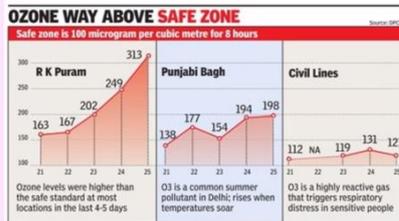
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## Ozone Pollution Worsens as Daily Temperatures Increases

Heat pulls city out of comfort ozone



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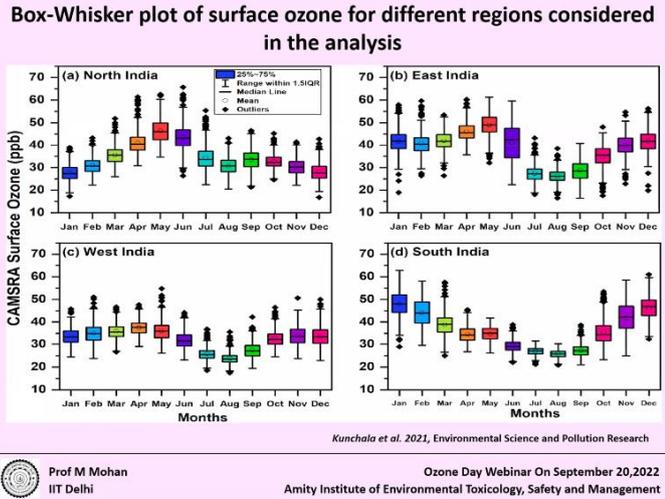
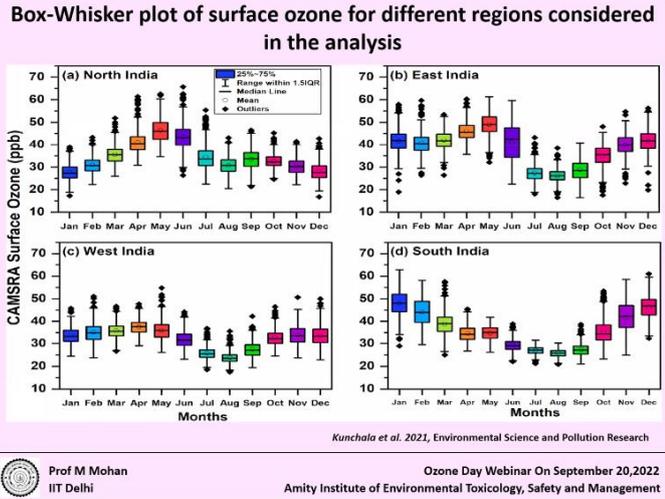


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**BEFORE COVID-19**



India

**DURING COVID-19**



January 8, 2020



Kangra valley in Dharmsala, India

Dhauladhar range is clearly visible during lockdown in Dharmsala



Source: <https://www.insider.com/before-after-photos-show-less-air-pollution-during-pandemic-lockdown>



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CITY	BEFORE vs. DURING LOCKDOWN* 2020					MARCH-APRIL 2019 vs. MARCH-APRIL 2020				
	PM <sub>2.5</sub>	PM <sub>10</sub>	NO <sub>2</sub>	CO	O <sub>3</sub>	PM <sub>2.5</sub>	PM <sub>10</sub>	NO <sub>2</sub>	CO	O <sub>3</sub>
Delhi	-41%	-52%	-50%	-29%	7%	-45%	-52%	-48%	-41%	-14%
Mumbai	-33%	-47%	-75%	-46%	8%	DNA*	DNA*	DNA*	DNA*	DNA*
Chennai	-14%	DNA*	-32%	-25%	3%	39%	DNA	-43%	-23%	73%
Bangalore	-22%	-34%	-60%	-16%	-11%	-47%	-40%	-56%	-15%	-21%
Kolkata	-23%	-34%	-60%	-29%	17%	-27%	-32%	-66%	-16%	87%

\* DNA: Data Not Available.

> 60%
40%-60%
20%-40%
< 20%

Decline in air pollutants

> 60%
40%-60%
20%-40%
< 20%

Increase in air pollutants

Percentage decline or increase in criteria air pollutants concentrations in five megacities of India during lockdown (25<sup>th</sup> March to 6<sup>th</sup> April 2020) and before lockdown (10<sup>th</sup> to 20<sup>th</sup> March 2020) period.

Source: Jain, S., & Sharma, T. (2020). Aerosol and Air Quality Research. <https://doi.org/10.4209/aaqr.2020.04.0171>



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### Processes Controlling Tropospheric Ozone are many and complex

Strat.-Trop. Exchange

OH

NMHCs, CH<sub>4</sub>, CO → HO<sub>2</sub>, RO<sub>2</sub>

NO → NO<sub>2</sub> → O<sub>3</sub> → NO

CO, O<sub>3</sub> → HO<sub>2</sub> → O<sub>3</sub> → HO<sub>2</sub>

OH

h<sub>v</sub>

H<sub>2</sub>O

Deposition

Emissions

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### Ozone Pollution Worsens as Daily Temperatures Increase

- Based on observed data from New York City for May to October ("smog season," averaging period not specified) for the years 1988 to 1990, and observed data (using a one-hour average) from Nashua, NH, for the years 2005 to 2010
- Both scatter plots show that the higher the temperature, the higher the ozone level, regardless of a city's size.
- Climate change is projected to bring higher average temperatures over this century, which could increase the occurrence of elevated ozone concentrations.

Source: Perera & Sanford, 2011 (data source: NAST 2001 (NYC); New Hampshire Department of Environmental Services 2011)

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### LULC over NCR region of India for 1972-2014

Urban and Built up
Water bodies
Shrublands
Dry Croplands
Mixed Croplands
Irrigated Croplands
Barren or Sparse vegetation
Mixed Forest

Ref. Sati and Mohan, *TsAC*, 2017

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### Experiment Design

Physics Options	
Microphysics	Lin
Cumulus	Kain Fritsch
Shortwave radiation	Goddard
Longwave radiation	RRTM
PBL scheme	YSU
Land surface model	Noah
Surface Physics	MM5
Chemistry Options	
Gas Phase mechanism	RACM
	CBMZ
Aerosol mechanism	MADE/SORGAM
	MOSAIC

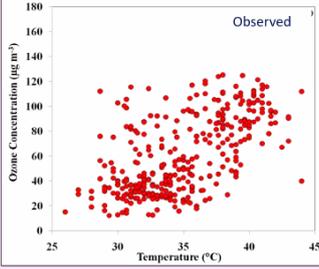
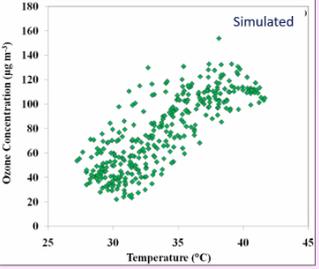
- ❖ Emission data: EDGAR monthly emission inventory
- ❖ Simulation Period: 1-15 June 2007, 2008, 2009 and 2010

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### Effect of Temperature on Ozone Production

\*Simulated Ozone concentration is always above 90  $\mu\text{g m}^{-3}$  for the temperature of 38°C and above. For the corresponding observed values, Ozone concentration is between 40  $\mu\text{g m}^{-3}$  to 120  $\mu\text{g m}^{-3}$  for the temperatures above 38°C.

\*When the temperature is as low as 28°C, the observed ozone concentration value is less than 30  $\mu\text{g m}^{-3}$  while simulated ozone concentrations range between 30  $\mu\text{g m}^{-3}$  to 70  $\mu\text{g m}^{-3}$ .

[Gupta and Mohan; Atmos. Environ., \(2015\), 122, 220-229](#)



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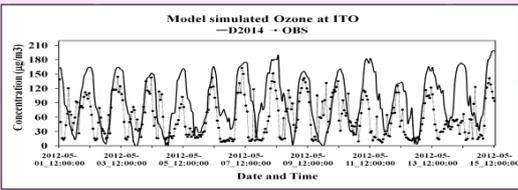
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### Model Validation- Case study May 2012



Statistical analysis of various pollutants averaged for the stations in Delhi				
Parameters	MNB (0)	MNE (0)	RMSE (0)	IOA (1)
Ozone	0.37	0.48	44.7	0.75
Reported Values	0.16	0.28	39.7	0.45
NO <sub>x</sub>	-0.51	0.54	46.9	0.85
Reported Values	-0.46	0.68		
SO <sub>x</sub>	-0.80	0.80	16.9	0.30
Reported Values	-0.23	0.76		
CO	-0.75	0.75	1.2	0.54
Reported Values	-0.60	0.65	1.0	0.60

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## Changes in the UHI intensities between D2014-D1972

### Surface UHI

Night Time  
1:30 IST

Increase (K)	Area (sq Km)
6+	9
5-6	667
4-5	794
3-4	68
1-3	80

- For nocturnal SHI most of the area experience an increase of 4-5 °C followed closely by 5-6 °C rise.

Mohan et al., 2021; Urban Climate

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## Changes in Ozone concentration

D1993-D1972

D2003-D1972

D2014-D1972

The nocturnal ozone concentration at surface shows considerable increase (40%) with increase in urban expansion

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### Changes in Ozone concentration

**O3 and Temperature averaged for stations in Delhi**

**O3 and NOx averaged for stations in Delhi**

The oxides of nitrogen referred as NOx, play important roles in atmospheric chemistry producing tropospheric ozone.

**Daytime Chemistry**

$\text{NO}_2 + h\nu \rightarrow \text{NO} + \text{O}$  Equation 1

$\text{NO} + \text{O}_3 + \text{M} \rightarrow \text{O}_3 + \text{M}$  Equation 2

**Night time Chemistry**

$\text{NO} + \text{O}_3 \rightarrow \text{NO}_2 + \text{O}_2$  Equation 3

- Ozone chemistry is quite consistent with the temperatures where high temperature favor increased ozone production during daytime (Gupta and Mohan, 2015, Im et al., 2011).
- NOx plays an important role in nocturnal O<sub>3</sub> chemistry and Awang et al., (2015) showed that low nighttime NOx concentration evidently decreased O<sub>3</sub> removal rates, thereby allowing O<sub>3</sub> to remain at high concentration at night.

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Global ranking of risk factors by total number of deaths from all causes for all ages and both sexes in 2016

Ref: State of Global Air 2018. Health Effects Institute

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## 5. Prof. Umesh Kulshrestha- “How Continuous O<sub>3</sub> Monitoring without Chlorine Measurements is Incomplete”

5.1. There is 95 % certainty that human activities are responsible for global warming.

5.2. Carbon dioxide is at an unprecedented level not seen for at least the last 8000,000 years.

- 5.3. Sea level is set to continue to rise at a faster rate than over the past 40 years.
- 5.4. Over the last two decades, the Greenland and Antarctic ice sheets have been melting and glaciers have been recorded in most parts of the world.
- 5.5. The ban on the burning of plastic wastes and other polymers reduced emissions of chlorine species.
- 5.6. In addition, NO emission was very less due to shut down of industries, flights, automobiles, etc.
- 5.7. Regular monitoring of O<sub>3</sub> is required.
- 5.8. Monitoring of HCl and Cl<sub>2</sub> is required.
- 5.9. Role of sanitizer studied is needed.
- 5.10. Role and interference of atmospheric dust to be investigated.
- 5.11. Policies are needed to support more research.

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75 Azadi Ka Amrit Mahotsav

Dr. Umesh Kulshrestha  
Professor  
School of Environmental Sciences, JNU, Delhi

Dr. Kulshrestha is currently a professor at School of Environmental Sciences, Jawaharlal Nehru University, New Delhi. He is also Honorary Director, Trans-Disciplinary Research Cluster, JNU and Honorary Deputy Director at South Asia Nitrogen Center. Dr Kulshrestha earned his PhD from Dayalbagh Educational Institute, Agra. He has a double post doc from National Physical Laboratory, New Delhi and Dept. of Meteorology, Stockholm University, Sweden. His research interests include Air Pollution, Atmospheric Chemistry and Climate Change, Environmental Analytical Chemistry, Environmental Impact Assessment. among many others. He is a fellow of Indian Geophysical Union (IGU) and has received many awards and recognitions. He has numerous research publications to his credit.

AMITY UNIVERSITY  
WORLD OZONE DAY CELEBRATION  
September 20, 2022  
03:00 to 05:00 PM  
Ms. Jayati Arora

The theme for 2022 "Multiscale pathways to 35% global reduction in life cycle emissions"

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## How Continuous O<sub>3</sub> Monitoring without Chlorine Measurements is Incomplete

For Source Identification in Urban Areas?



**Umesh Kulshrestha**  
 Professor  
 Deputy Director, South Asian Nitrogen Centre  
 Former Member Member- EPCA  
 School of Environmental Sciences  
 Jawaharlal Nehru University  
 New Delhi  
 Email: umeshkulshrestha@gmail.com

JNU

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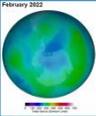
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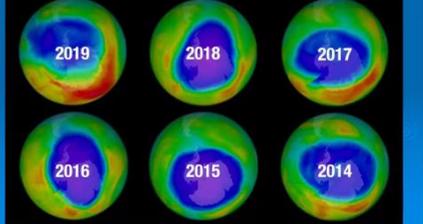
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## Ozone hole is healed in 2020

February 2020



Purple colour (below 220 DU) disappears



2019 2018 2017  
 2016 2015 2014

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### Ozone hole is healed in 2020

February 2022

Purple colour (below 220 DU) disappears

2019 2018 2017  
2016 2015 2014

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### COVID 19 Lockdown Scenario

Major anthropogenic activities are stopped

Emissions of CO<sub>2</sub> are reduced drastically

Industries Transport

Monthly CO<sub>2</sub> Concentration (ppm)

As of March 2020, little or no atmospheric evidence yet of reduced global CO<sub>2</sub> emissions from COVID-19 Economic Slowdown

### Major Findings of IPCC (2013)

- 1 There is 95 percent certainty that human activities are responsible for global warming
- 2 Carbon dioxide is at an "unprecedented" level not seen for at least the last 800,000 years
- 3 Sea level is set to continue to rise at a faster rate than over the past 40 years
- 4 Over the last two decades, the Greenland and Antarctic ice sheets have been melting and glaciers have receded in most parts of the world

\* IPCC Assessment Report Summary for Policy Makers, released Sept. 27, 2013  
http://www.ipcc.ch/

Carbon Dioxide Variations

The instrumental recordation from Mauna Loa Observatory has revealed a dramatic rise in CO<sub>2</sub>

CO<sub>2</sub> Concentration (ppm)

Thousands of Years Ago

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## Increased Ambient Ozone in Delhi

Month	2017-18 (ug/m³)	2018-19 (ug/m³)
Sept	18	18
Oct	35	35
Nov	60	60
Dec	60	60
Jan	42	42
Feb	35	35
Mar	40	40
Apr	10	52
May	10	58
Jun	10	65
Jul	10	38
Aug	18	48

Kulshrestha and Mishra, 2019

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## Ozone at DTU, Delhi (CPCB)

From CPCB (<https://app.cpcbcr.com/cpr/#/caaqm-dashboard-all/caaqm-landing>)  
Data retrieved on 16/03/2020

Effects of the following factors to be studied further:

- HCl/Cl<sub>2</sub> quenching
- NO quenching
- Black carbon dimming
- VOCs/Sanitizer effect

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### Ozone at DTU, Delhi (CPCB)

From CPCB (<https://app.cpcbcr.com/crr/#/caaqm-dashboard-all/caaqm-landing>)  
Data retrieved on 16/09/2020

Effects of the following factors to be studied further:

- HCl/O<sub>3</sub> quenching
- NO quenching
- Black carbon dimming
- VOCs/Sanitizer effect

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## Stratospheric Ozone

How the ozone layer is formed and maintained?  
**The Chapman Cycle: 1930 by Sydney Chapman**  
3 steps and 4 chemical reactions

Step 1. Initiation  $O_2 + h\nu \rightarrow O + O$  (120 – 210 nm) ..... (i)

Step 2. Propagation (cycling)

$O + O_2 + M \rightarrow O_3 + M^*$  (generates heat) ..... (ii)

$O_3 + h\nu \rightarrow O_2 + O$  (220 – 320 nm) ..... (iii)

Step 3. Termination  $O_3 + O \rightarrow 2O_2$  ..... (iv)

Dissociation product of O<sub>3</sub> depends on wavelength:

At 175nm  $O_3 + h\nu \rightarrow O(^1D) + O(^3P)$

At 242 nm  $O_3 + h\nu \rightarrow O(^1P) + O(^3P)$

At 133.2 nm  $O_3 + h\nu \rightarrow O(^3P) + O(^1S)$

## Tropospheric Ozone

**Ozone from NO<sub>2</sub>**  
NO<sub>2</sub> + radiations (>380 nm) → NO + O  
O + O<sub>2</sub> → O<sub>3</sub>

**Ozone from carbon monoxide**  
CO + 2O<sub>2</sub> + hν → CO<sub>2</sub> + O<sub>3</sub>

**Ozone from methane**  
CH<sub>4</sub> + 4O<sub>2</sub> + 2hν → HCHO + H<sub>2</sub>O + 2O<sub>3</sub>  
HCHO + hν → H + HCO (λ < 330nm)  
HCO + hν → H + CO (λ < 360nm)  
CO + 2O<sub>2</sub> + hν → CO<sub>2</sub> + O<sub>3</sub>

**Ozone from non methane hydrocarbons**  
RH + 4O<sub>2</sub> + 2hν → RCHO + H<sub>2</sub>O + 2O<sub>3</sub>

Source reactions of NO<sub>2</sub>  
2 NO + O<sub>2</sub> → 2 NO<sub>2</sub>

Occurs in power plant plumes and vehicular tail pipes with very high [NO]

NO + HO<sub>2</sub> → OH + NO<sub>2</sub>  
NO + RO<sub>2</sub> → RO + NO<sub>2</sub>

Occurs at natural levels of [NO]

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The ban on burning of plastic wastes and other polymers reduced emissions of chlorine species.

The HCl consumes  $O_3$ .

The net  $O_3$  levels were relatively low in early 2019 and before.

At present, in the absence of HCl emissions post ban, the ozone levels were built up.

In addition, NO emissions were very less due to shut down of industries, flights, automobiles etc.

As we know that similar to HCl, the NO also destructs ozone-

Hence, the present limited HCl/ $Cl_2$  and NO atmospheric conditions are responsible for building up of  $O_3$  levels in the NCR-Delhi and such urban locations.

The rural areas have more  $NO_2$  than NO and in Delhi air similar to rural areas.

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## 6. Prof. Qing-Bin Lu- “Success and New Implications of the Montreal Protocol”

6.1. The Montreal Protocol is the most successful international treaty.

6.2. CFCs are not only major ozone-depleting gases but powerful greenhouse gases.

6.3. CFCs played a major role in global warming.

6.4. Cosmic-Ray theory of the ozone hole was given in 1998.

6.5. Cosmic rays could be a major contributor to ozone destruction over Antarctica.

6.6. Photolysis by UV photons, dissociative electron attachment by free electrons, and dissociative electron transfer reaction are the possible pathways for the destruction of CFCs in the atmosphere.

6.7. CFC theory of global warming was given in 2009.

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**Prof. Qing-Bin Lu**  
 Professor; Undergraduate  
 Advisor - Chemical Physics  
 University of Waterloo, Ontario,  
 Canada

Dr. Lu (PI) is a Professor in Physics, with cross appointments to the Departments of Biology and Chemistry at U of Waterloo. Holding a CIHR Senior Research Fellowship, he received training in radiation chemistry and radiation biology at U of Sherbrooke with Dr. Leon Sanche, and in femtochemistry and femtobiology at Caltech with Nobel laureate Dr. Ahmed Zewail. Dr. Lu's research at Waterloo has mainly focused on theories of the ozone hole and global climate change and on femtomedicine and cancer therapies. His main contributions include discoveries of a dissociative electron transfer (DET) reaction mechanism for various molecular systems of atmospheric, environmental and biological significance, ranging from atmospheric ozone depletion to the activation of anticancer drugs and reductive DNA damage, discoveries/inventions of novel cancer therapies (16 issued patents), discoveries and development of a new theory of the ozone hole and a new theory of modern global warming, and discovery of the tropical ozone hole. He has published many reputed research papers, and received awards like Early Researcher Award, The Ontario Ministry of Research and Innovation (MRI), New Investigator Award, Canadian Institutes of Health Research (CIHR), University Research Chair, University of Waterloo.



**Ms. Charu Singh**

AMITY UNIVERSITY  
 WORLD OZONE DAY CELEBRATION  
 September 20, 2022  
 03:00 to 05:00 PM  
 Ms. Charu Singh

The theme for 2022 "Montreal Protocol: 35: global cooperation protecting life on earth"

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**Ms. Jayati Arora**



**Prof. Tanu Jindal**



**Prof. Qing-Bin Lu**



**Ms. Charu Singh**



**Prof. Qing-Bin Lu**

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### Ozone Depletion without Polar Stratospheric Clouds (PSCs): Gas-Phase Reaction

CFCs  
+  
**Sunlight**

↓

**Chlorine Atoms**

↓

Destroy Ozone Molecules

$$\text{F}-\overset{\text{Cl}}{\underset{\text{F}}{\text{C}}}-\text{Cl} \xrightarrow{+\text{UV light}} \text{F}-\overset{\text{Cl}}{\underset{\text{F}}{\text{C}}}-\text{F} + \text{Cl}$$

**1 Cl atom can destroy 100,000 O<sub>3</sub> molecules!**

R. Stolarski & R. J. Cicerone, Can. J. Chem. 52, 1610-1615 (1974).  
M. J. Molina & F. S. Rowland, Nature 249, 810-812 (1974).

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### Cl Atom Production Rate from CFC Photodissociation and Ozone Hole

**Cl production rate (cm<sup>-3</sup> sec<sup>-1</sup>)**

Chlorine from Photodissociation of CFCs

**Ozone Loss (%)**

Antarctic Ozone Hole

**Altitude (km)**

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## Ozone Depletion with Polar Stratospheric Clouds (PSCs): Surface Reactions

**In winter darkness**

- Formation of polar stratospheric clouds:
 
$$\text{HNO}_3 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{O} \cdot \text{HNO}_3$$
- Activation of
 
$$\text{ClONO}_2 + \text{H}_2\text{O} \rightarrow \text{HCl} + \text{ClONO}$$

**In the light of the rising spring sun**

- Formation of reactive Cl atom:
 
$$\text{Cl}_2 + h\nu \rightarrow 2\text{Cl}$$
- Ozone depleting reaction cycles:
 
$$\begin{aligned} \text{Cl} + \text{O}_3 &\rightarrow \text{ClO} + \text{O}_2 \\ \text{ClO} + \text{ClO} &\rightarrow \text{Cl}_2\text{O}_2 \\ \text{Cl}_2\text{O}_2 + h\nu &\rightarrow 2\text{Cl} + \text{O}_2 \\ \text{Cl} + \text{O}_3 &\rightarrow \text{ClO} + \text{O}_2 \end{aligned}$$

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## Montreal Protocol signed on September 16, 1987

**Montreal Protocol**

Antarctic

Mid-Latitudes

1980 stratospheric levels

Year

Equivalent Effective Stratospheric Chlorine (ppt)

Legend:

- NOAA obs. weighted trop. data
- WMO scenario. weighted trop.
- EESC(A)
- NOAA obs. weighted trop. data
- WMO scenario. weighted trop.
- EESC(ML)

(Credits: NOAA's Global Monitoring Laboratory)

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## Cosmic-Ray Theory of the Ozone Hole (1998--)

Lu & Madey, *J. Chem. Phys.* 111, 2861(1999).  
 Lu & Sanche, *Phys. Rev. Lett.* 87, 078501(2001).  
 Lu, *Phys. Rev. Lett.* 102, 118501(2009).  
 Lu, *Phys. Rep.* 487, 141-167(2010).  
 Lu, *Int. J. Mod. Phys. B*27, 1350073(2013).  
 Qing-Bin Lu, "New Theories and Predictions of the Ozone Hole and Climate Change" (World Scientific, 1<sup>st</sup> Ed., 2015; 2<sup>nd</sup> Ed., 2023).  
 Lu, *AIP Advances* 11, 115307 (2021).  
 Lu, *AIP Advances* 12, 075006 (2022).  
 Lu, to be published.

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## Possible Pathways for Destruction of CFCs in the Atmosphere

- 1) Photolysis by UV photons
 
$$\text{CF}_2\text{Cl}_2 + h\nu \rightarrow \text{CF}_2\text{Cl} + \text{Cl}$$
 (Dissociation cross section  $\sim 10^{-20} \text{ cm}^2$ )
- 2) Dissociative electron attachment (DEA) by free electrons
 
$$\text{CF}_2\text{Cl}_2 + (\sim 0\text{eV } e^-) \rightarrow \text{CF}_2\text{Cl} + \text{Cl}^-$$
 (Dissociation cross section  $\sim 10^{-16} \text{ cm}^2$ )
- 3) Dissociative electron transfer (DET) reaction
 
$$\text{CF}_2\text{Cl}_2 + e^- (\text{H}_2\text{O})_n \rightarrow \text{CF}_2\text{Cl} + (\text{H}_2\text{O})_n + \text{Cl}^- \text{ (on Ice)}$$
 (Dissociation cross section is  $10^{-14} \sim 10^{-12} \text{ cm}^2$ )

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## Fingerprints of the CRE mechanism

Lu, AIP Advances 11, 115307 (2021).

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## Success of the Montreal Protocol (I)

A Antarctic (60-90° S) Oct-Dec Mean  
 $R^2=0.71$

B Near-Global (60°S-60°N) Annual Mean  
 $R^2=0.79$

Lu, Int. J. Mod. Phys. B27, 1350073(2013).  
 Qing-Bin Lu, "New Theories and Predictions of the Ozone Hole and Climate Change" (World Scientific, 1<sup>st</sup> Ed., 2015; 2<sup>nd</sup> Ed., 2023).  
 Lu, AIP Advances 11, 115307 (2021).

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## Discovery of the tropical ozone hole

1980s minus 1960s O<sub>3</sub> Change (%) SON

2000s minus 1980s O<sub>3</sub> Change (%) SON

1990s minus 1960s O<sub>3</sub> Change (%) SON

2010s minus 1980s O<sub>3</sub> Change (%) SON

Lu, AIP Adv. 12, 075006 (2022).

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## Success of the Montreal Protocol (II)

(a)-(c) Ozone loss (%)  
 (a) Antarctica 60°S-90°S, SCIN  
 (b) Tropics 30°N-30°S, Annual  
 (c) Arctic 60°N-90°N, MAM  
 (d) 1980s minus 1960s O<sub>3</sub> Change (% Annual)  
 (e) 2000s minus 1980s O<sub>3</sub> Change (% Annual)  
 (f) Temperature (K), 2002-2010 RCM GAT

Lu, AIP Adv. 12, 075006 (2022).

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## Tropical ozone hole (continued)

Ozone change (psv), 1980s wrt pre-1980, 13.5-20.5 km  
 Ozone Change (%), 1980s wrt pre-1980, 13.5-20.5 km  
 Ozone Change (psv), 1990s wrt pre-1980, 13.5-20.5 km  
 Ozone Change (%), 1990s wrt pre-1980, 13.5-20.5 km

QB Lu, to be published

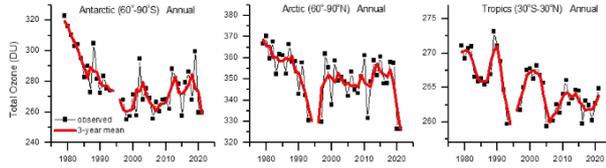
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## Why do we worry about the tropical ozone hole?



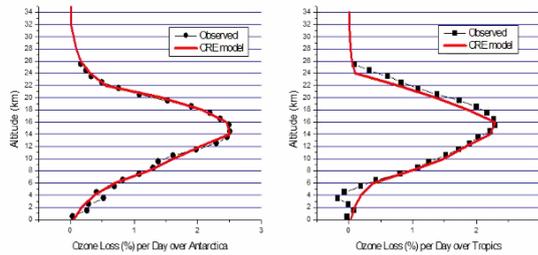
As satellite data show, the annual average total ozone value over the tropics was around 262 DU over the last decade, while it was 275 DU over Antarctica and 288 over the global!

QB Lu, to be published



Prof. Qing-Bin Lu

## Observation and Theoretical calculations of global ozone depletion



QB Lu, to be published



Prof. Qing-Bin Lu

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**CFCs are the Major Culprit of Global Warming**

Global surface temp change  $\Delta T_s$  ( $^{\circ}\text{C}$ )

CO<sub>2</sub>-warming

CFC-Warming

Cooling

— ENSO & volcanic effects removed  
— Russia and Alaska excluded

The global warming was mainly caused by the greenhouse effect of CFCs rather than CO<sub>2</sub> – Success of the Montreal Protocol (III)!

Lu, Atmosphere 13(9), 1419 (2022).

Prof. Qing-Bin Lu

Vote of thanks

The vote of thanks was given by Ms. Prangya Rath.

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Prof. Tanu Jindal

Ms. Charu Singh

Ms. Jayati Arora

Ms. Yogyata Srivastava

Ms. Prangya Rath

